## phys522: HW #2

1. The probability density is for the hydrogen atom  $\rho_{n\ell m} = |\psi_{n\ell m}|^2$  and the only part of the wave function that is imaginary comes from the  $\phi$  dependence. We can therefore write the wave function as:

$$\psi_{n\ell m}(\vec{r}) = |\psi_{n\ell m}(\vec{r})|e^{im\phi} = \sqrt{\rho_{n\ell m}(\vec{r})}e^{im\phi}$$

Show that the probability current is proportional to the quantum number m and given by

$$\vec{J}_{n\ell m} = m \frac{\hbar}{\mu} \frac{\rho_{n\ell m}(\vec{r})}{r \sin \theta} \hat{e}_{\phi}$$

Comment on the phrase "stationary states".

2. Commins 8.6

8.6. (a) Consider a bound-state wave function  $\psi(\mathbf{r})$  with corresponding probability density  $\rho = \psi^* \psi$  and probability current density

$$\boldsymbol{j} = \frac{\hbar}{2mi} (\boldsymbol{\psi}^* \nabla \boldsymbol{\psi} - \boldsymbol{\psi} \nabla \boldsymbol{\psi}^*)$$

Assume that  $\rho$  satisfies the condition

$$\int \rho \mathbf{r} \times \hat{\mathbf{n}} \, ds = 0 \tag{1}$$

where the quantity on the left-hand side of (1) is a surface integral taken over a surface at  $\infty$ , and  $\hat{n}$  is a unit normal vector to the surface. Show that the expectation value of the orbital angular-momentum operator L is

$$\langle \boldsymbol{L} \rangle = \boldsymbol{m} \int \boldsymbol{r} \times \boldsymbol{j} \, d\tau \tag{2}$$

where the integral on the right-hand side of (2) is a volume integral taken over all space. (b) Now, considering the description of a particle with spin-1/2 in nonrelativistic quantum mechnics, let us assume that  $\psi = \phi(r)\chi$ , where  $\phi$  is a bounded spatial wave function and  $\chi$  is a two-component spinor. We write the expectation value of the spin operator as

$$\langle \boldsymbol{S} \rangle = \frac{\hbar}{2} \int \boldsymbol{\psi}^* \boldsymbol{\sigma} \boldsymbol{\psi} \quad d\tau \tag{3}$$

Show that

$$\psi^* \boldsymbol{\sigma} \boldsymbol{\psi} = \frac{1}{2} \boldsymbol{r} \times \left[ \nabla \times (\psi^* \boldsymbol{\sigma} \boldsymbol{\psi}) \right] - \frac{1}{2} \nabla \left[ \boldsymbol{r} \cdot \boldsymbol{\psi}^* \boldsymbol{\sigma} \boldsymbol{\psi} \right] + \frac{1}{2} \sum_{i=1}^3 \frac{\partial}{\partial x_i} \left[ x_i \left( \psi^* \boldsymbol{\sigma} \boldsymbol{\psi} \right) \right]$$
(4)

and show that the last two terms on the right-hand side of (4) make no contribution to the integral in (3). Thus, in analogy to (2), (3) can be written as

$$\langle \boldsymbol{S} \rangle = \boldsymbol{m} \int \boldsymbol{r} \times \boldsymbol{j}_{\boldsymbol{S}} \, d\,\boldsymbol{\tau} \tag{5}$$

Here  $\mathbf{j}_{S}$ , the spin current density, is  $\mathbf{j}_{S} = \nabla \times \mathbf{V}_{S}$ , where

$$V_{S} = \frac{\hbar}{4m} \psi^{*} \boldsymbol{\sigma} \psi \tag{6}$$

(c) Calculate the spin current density in atomic units for the  $2^2s_{1/2}$ ,  $m_s = 1/2$  state of atomic hydrogen.