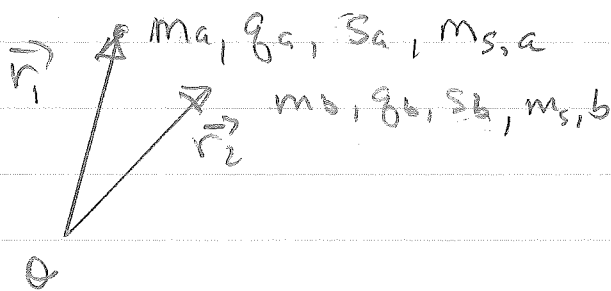


Lec 14: Identical Particles

QM insists that multiparticle wave function not distinguish identical particles

Two Particle wave function:



Distinguishable: by mass or charge

$$\Psi_{ab}(1,2) = \Psi_a(\vec{r}_1) \chi_a^{(1)} \Psi_b(\vec{r}_2) \chi_b^{(2)}$$

Indistinguishable (identical) same mass, charge
(probability density function)

P.D.F. Ψ^* may not allow a measurement to distinguish them. So P.D.F. must be symmetric under particle exchange — exchange symmetry

$$|\Psi_{ab}(1,2)|^2 = |\Psi_{ab}(2,1)|^2$$

leaves two possibilities:

$$\Psi_{ab}(1,2) = \pm \Psi_{ab}(2,1)$$

Spin-Statistics theorem of Quantum Field Theory
and experimental fact:

$$\Psi_{ab}(1,2) = +\Psi_{ab}(2,1) \quad \text{integer-spin "Bosons"}$$

$$\Psi_{ab}(1,2) = -\Psi_{ab}(2,1) \quad \frac{1}{2} \text{ integer-spin "Fermions"}$$

Strikingly different statistical behavior:
Bose-Einstein, Fermi-Dirac

Pauli Exclusion Principle: No two electrons
can occupy the same state

$$\Psi_{ab} = \frac{1}{\sqrt{2}} \left[\Psi_a(\vec{r}_1) \chi_a^{(1)} \Psi_b(\vec{r}_2) \chi_b^{(2)} - \Psi_a(\vec{r}_2) \chi_a^{(2)} \Psi_b(\vec{r}_1) \chi_b^{(1)} \right]$$

explicitly zero if $a=b$.

How to build anti-symmetrized two electron
wave function.

$$\begin{aligned} \Psi_{BT}(1,2) &= \Psi_{\text{space}}(\vec{r}_1, \vec{r}_2) \chi_{\text{spin}} \\ &= \begin{cases} \Psi_{\text{space}}^A \chi^S & A \equiv \text{anti-symmetric} \\ \Psi_{\text{space}}^S \chi^A & S \equiv \text{symmetric} \end{cases} \end{aligned}$$

General rule for combining angular momentum j_1 and j_2 :

$$|j_1 - j_2| \leq j \leq j_1 + j_2$$

Example, 2 spins $\frac{1}{2}$:

$$|\frac{1}{2} - \frac{1}{2}| \leq j \leq \frac{1}{2} + \frac{1}{2} \Rightarrow j = 0, 1$$

In group-theoretic notation, multiplets are denoted by degeneracy.

$$\underline{2} \otimes \underline{2} = \underline{1} \oplus \underline{3}$$

explicitly shows sum of states is the same

Spinor states $\chi(s, m_s)$

notation: $\chi(\frac{1}{2}, \frac{1}{2}) \chi(\frac{1}{2}, -\frac{1}{2}) \equiv \uparrow \downarrow$
order indicators 1, 2

$$\underline{2} \otimes \underline{2} : \uparrow\uparrow, \uparrow\downarrow, \downarrow\uparrow, \downarrow\downarrow$$

$$\underline{3} : \begin{cases} \chi(1, 1) = \uparrow\uparrow \\ \chi(1, 0) = \frac{1}{\sqrt{2}} (\uparrow\downarrow + \downarrow\uparrow) \\ \chi(1, -1) = \downarrow\downarrow \end{cases} \text{ symmetric}$$

$$\underline{1} : \chi(0, 0) = \frac{1}{\sqrt{2}} (\uparrow\downarrow - \downarrow\uparrow) \text{ antisymmetric}$$

Group theoretic multiplets (irreducible representations) always have a definite symmetry.

Irreducible - under rotations, components of multiplets mix, but not between multiplets

example - 3 quarks

$$\underline{2} \otimes \underline{2} \otimes \underline{2} = (\underline{1} \oplus \underline{3}) \otimes \underline{2}_3$$

$$= \underline{2}_{\{1,2\}} \oplus \underline{2}_{\{2,3\}} \oplus \underline{4}_{\{1,2,3\}}$$

antisymmetric

spin $\frac{1}{2}$ baryons
p, n

↑ symmetrized
S = $\frac{3}{2}$ baryons
Δ

Helium

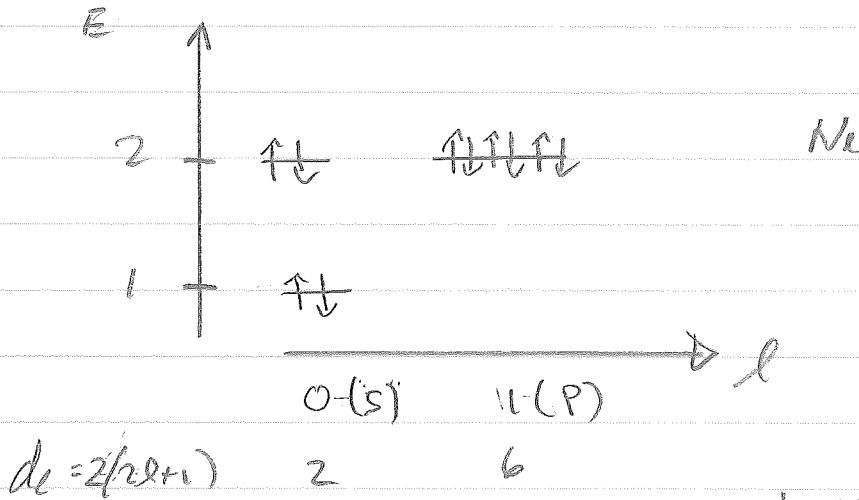
$$\psi_{\text{ground}} = \psi^S \chi^A(0,0)$$

$$\psi^S = \psi_{100}(\vec{r}_1) \psi_{100}(\vec{r}_2)$$

$$\chi_0 = \frac{1}{\sqrt{2}} (\uparrow\downarrow - \downarrow\uparrow)$$

Multi electron atoms - Shell model

electrons fill shells according to hydrogen principle quantum number n .



$d_n = 2n^2$

- H 1s¹ ← #electrons
- He 1s² filled shell
- Li 1s² 2s¹ ...
- Ne 1s² 2s² 2p⁶

Order of filling:

see for example,

<http://hyperphysics.phy-astr.gsu.edu/hbase/pertab/perfill.html#c2>

