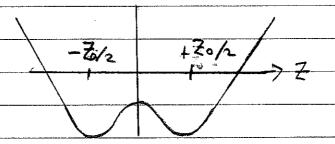
Lecture 16: Covalent Bond

Approximate proton separation as fixed.
Potential seen by electron is double-well:

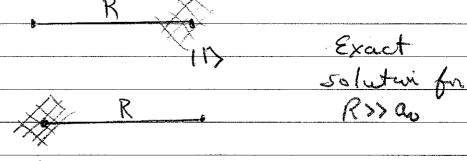


Constant proton separation 20 juilified by slow proton oscillations

So expect wp ~ mp w

Coordinate System:

	perturbative	approach	1	H +	P
-					-1-



2>

$$\langle F^{2}|1^{2}\rangle = \frac{1}{\sqrt{\pi a_{0}^{3}}} e^{-\frac{1}{2}/a_{0}}$$
 degenerate
$$\langle F_{2}^{2}|2^{2}\rangle = \frac{1}{\sqrt{\pi a_{0}^{3}}} e^{-\frac{1}{2}/a_{0}}$$
 state

$$\hat{H} = \frac{\hat{e}^2}{2m} - \frac{e^2}{r_1} - \frac{e^2}{r_2} + \frac{e^2}{R}$$

H matrix:

Where -A in energy due to turneley across potential barrier Separating States 117, 127.

es heturi :

We have assumed (1/2)=0 which is not correct. In troduce overlap

get (see Hw)

$$F_{I} = \frac{1}{110} \left(E_{0} - A \right) \cdot |II\rangle = \frac{1}{\sqrt{2+20}} \left(|I| + |I| \right)$$

& Further lower the ground state energy Following Pauling and Wilson,

with E, = H-atom ground state energy

and

off diagonal (turneleng) terms are

giving
$$A = K - \delta(E, +\frac{e^2}{R})$$

then in terms of J, K

$$= E_1 + \frac{e^2}{R} - \left(\frac{J+K}{I+\Delta}\right)$$

$$E_{\Gamma} = E_{1} + \frac{e^{2}}{R} - \left(\frac{J-K}{I-D}\right)$$

for completeness, integrile are $(D \equiv \frac{R}{a_0})$

$$\overline{J} = \frac{e^2}{a_0} \left[-\frac{1}{D} + e^{-2D} \left(1 + \frac{1}{D} \right) \right]$$

	Ex in then a function of R and have	
	a minimum @ R = 0,13 nm and value	-15.4eV
b	nding or dissociation=pulling the molecule apart into proton and hyd	
	binding energy is - Epind = E - E,	
	after dissociation, electron is bound to one of the p	rotons
	Epind = (-13.6 eV) - (-15.4eV) = 1.8 eV	
	Exp valeur R = 0,106 nm Ebrid = 2	. 8eV
1.0.1.27		
	We have assumed wave furctions are	
	We have assumed wave furctions are undistorted by presence of other proton.	
	Covalent bond is due to tunveling between	Jeen
)	degenerate stater	
	O .	
2		-
- 2		
		,
J		